



Degenerate rearrangements of long-lived 9,10-dimethyl-9-(1-methyl-1-vinyl)-phenanthrenium and 9,10-dimethyl-9-(*trans*-1-methyl-1-propenyl)phenanthrenium ions: unexpected strong retardation of vinyl migrations

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DOI: 10.1070/MC2006v016n02ABEH002209

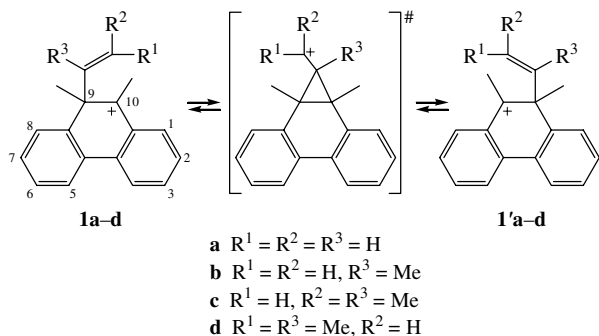
As found by dynamic NMR, the rates of degenerate rearrangements of long-lived 9,10-dimethyl-9-(1-methyl-1-vinyl)phenanthrenium and 9,10-dimethyl-9-(*trans*-1-methyl-1-propenyl)phenanthrenium ions proceeding by 1,2-vinyl shifts are unexpectedly much lower than those of the 9,10-dimethyl-9-vinylphenanthrenium and 9,10-dimethyl-9-(*cis*-1-methyl-1-propenyl)phenanthrenium ions.

Long-lived 9-vinyl-substituted 9,10-dimethylphenanthrenium cations capable of degenerate rearrangement by 1,2-vinyl shifts (Scheme 1) are suitable for determining the migration ability of vinyl and substituted vinyl groups.^{1,2} The migration ability of unsubstituted vinyl and *cis*-1-methyl-1-propenyl groups (ions **1a,c**) is the highest among hydrocarbon migrants, *cf.* refs. 3, 4.

We studied the kinetics of degenerate rearrangements of 9,10-dimethyl-9-(1-methyl-1-vinyl)- and 9,10-dimethyl-9-(*trans*-1-methyl-1-propenyl)phenanthrenium ions (**1b** and **1d**, respec-

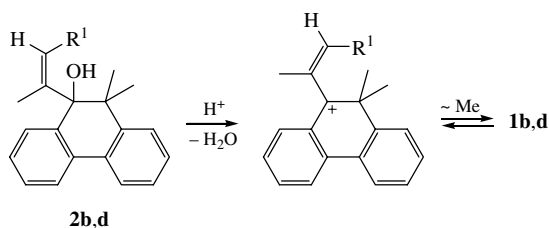
tively) proceeding by 1,2-shifts of vinyl groups by dynamic NMR. Cations **1b** and **1d** were generated under long-life conditions by the ionization of carbinols **2b,d**[†] in FSO₃H–SO₂ClF–CD₂Cl₂ at –130 °C (Scheme 2). Their structures were confirmed by ¹H and ¹³C NMR spectroscopy.[‡]

[†] Satisfactory NMR, IR and MS data were obtained for compounds **2b,d** prepared from 10,10-dimethyl-9-oxo-9,10-dihydrophenanthrene⁵ and CH(R)=C(Me)MgBr (R = H, Me) in THF.



Scheme 1

As the temperature increased, reversible changes in the 1H NMR spectra of cations **1b** and **1d** typical of degenerate rearrangements of the carbocations of such a type by 1,2-vinyl shifts were observed. The alternative mechanisms of similar degenerate



Scheme 2

‡ For **1b** (–107 °C): 1H NMR, δ : 8.87–8.52 (m, 4H), 8.05–7.64 (m, 4H), 5.89 (s, 1H), 5.77 (s, 1H), 3.42 (s, 3H), 1.77 (s, 3H), 1.22 (s, 3H). ^{13}C NMR, δ : 231.6 (s), 149.4 (s), 147.8 (s), 142.4 (s), 133.7 (s), 127.2 (s), 154.4 (d), 137.5 (d), 134.6 (d), 131.1 (d), 129.5 (d), 128.1 (d), 127.3 (d), 126.1 (d), 118.2 (t), 63.9 (s), 28.5 (q), 24.6 (q), 21.0 (q).

For **1d** (–114 °C): 1H NMR, δ : 9.0–8.4 (m, 4H), 8.2–7.7 (m, 4H), 5.70 (q, 1H, J 6 Hz), 3.40 (s, 3H), 2.50 (s, 3H), 1.72 (s, 3H), 0.57 (d, 3H, J 6 Hz). ^{13}C NMR, δ : 233.4 (s), 151.9 (s), 147.0 (s), 134.3 (s), 131.9 (s), 126.1 (s), 152.5 (d), 136.8 (d), 134.2 (d), 131.0 (d), 129.1 (d), 128.3 (d), 127.6 (d), 127.1 (d), 125.9 (d), 62.4 (s), 31.4 (q), 25.7 (q), 24.1 (q), 14.4 (q) (Bruker AM-400 spectrometer; CD_2Cl_2 as an internal standard, δ 5.33 for 1H and 53.6 for ^{13}C).

Table 1 Rates of 1,2-shifts of vinyl groups in cations **1a–d**.

Ion	k^a/s^{-1}	$T/^\circ C$	$\Delta G^\ddagger/kJ\ mol^{-1}$	k^b/s^{-1}
1a ²	1.1×10^5	–103	24	1×10^5
1b	12 ± 2	–102 \pm 2	37.5 ± 0.5	10 ± 4
1c ¹	1.0×10^5	–120	22	6×10^5
1d	40 ± 10	–88 \pm 2	38.9 ± 0.6	4 ± 2

^aExperimental values. ^bAt –103 °C, calculated from the Eyring equation.

rearrangements were thoroughly analysed and firmly rejected previously.^{1,2} The rates of the degenerate rearrangements of cations **1b** and **1d** by 1,2-shift of vinyl groups (Table 1) were determined by dynamic 1H NMR. Comparison of these data with those obtained earlier for cations **1a**² and **1c**¹ shows unexpected very strong retardation of 1,2-shifts of 1-methyl-1-vinyl and *trans*-1-methyl-1-propenyl groups. For example, in passing from cation **1c** having *cis*-1-methyl-1-propenyl group to **1d** having *trans*-1-methyl-1-propenyl group, the rate of rearrangement decreased by a factor of 10^5 . Note that the rate of migration of the *trans*-1-methyl-1-propenyl group is close to that of the 1-methyl-1-vinyl group in spite of the presence of a methyl group in the β -position of the former.

The reasons of the retarding effects observed are yet unknown. Now we try to solve this problem by quantum chemistry calculations.

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Received: 30th June 2005; Com. 05/2540